A New Triterpene and New Sesquiterpenes from the Roots of Ligularia sagitta

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One new triterpene and seven new eremophilane-type sesquiterpenes were isolated from the roots of Ligularia sagitta, among them $1a$, an O-acetylated oleanane type triterpene, $2a$ and $3a$, two 11-nor eremophilane-type sesquiterpenes, 4a, 5, and 6, three standard eremophilane-type sesquiterpenes, and 7 and 8, two tri-nor eremophilane-type sesquiterpenes. Their structures were established by extensive spectral analyses (1D, 2D-NMR, IR, and MS). Compound 1a showed moderate cytotoxicity against two tumor cell lines.

Introduction. – The genus Ligularia (Compositae) is widely distributed in China with about 100 species, of which more than 20 have long been used in traditional folk medicine. This genus possesses efficacies of antipyretics, relieving phlegm and cough, invigorating circulation of blood and soothing pain [1]. A previous study showed that the main components of Ligularia are eremophilane sesquiterpenes and pyrrolizidine alkloids with strong physiological activities [2]. In our long-standing interest in the study of biodiversity and searching for bioactive compounds from several *Ligularia* species [3], the results showed that the components from the same genus, even from the same species displayed remarkable taxonomic differences because of the variability in the ecological environments and collection seasons $[3c][4]$. As a part of our ongoing investigation on the chemical constitutents of the roots of Ligularia sagitta collected from Gannan Tibet Autonomous Region (2000 – 3800 m above sea level) in Gansu Province of P. R. China, one new triterpene and seven new eremophilane-type sesquiterpenes ($Fig. 1$) were obtained. In this report, we describe the isolation and the structure elucidation (by ¹H- and ¹³C-NMR, COSY, HMBC, and NOESY experiments) of these new compounds.

Results and Discussion. – 1. Structure Elucidation. The petroleum ether (b.p. 60 – 90°)/Et₂O/MeOH (1:1:1)-extracts of roots of L. sagitta were submitted to column chromatography on silica gel, yielding fractions that were further purified to give compounds $5-8$; compounds $1a-4a$ were obtained after acetylation of the corresponding mixture, otherwise, they were difficult to separate.

Compound 1a was obtained as a white amorphous powder. The HR-ESI-MS experiment gave an ion peak of $[M + NH_4]^+$ at m/z 616.4212, consistent with the molecular formula of $C_{36}H_{54}O_7(C_{36}H_{58}NO_7^+;$ calc. 616.4213), which accounted for ten degrees of unsaturation. The IR spectrum showed the absorption bands of $C=O(1737)$ and 1706 cm⁻¹) and C=C (1634 cm⁻¹).

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Fig. 1. Structures of compounds $1a-4a$ and $5-8$

The ${}^{1}H$ - and ${}^{13}C$ -NMR (*Table 1*) spectra of compound **1a** showed the typical signals of three AcO groups at $\delta(H)$ 1.94 (s, Me), 2.01 (s, Me), 2.09 (s, Me), and $\delta(C)$ 170.3 (C) , 170.3 (C) , 171.5 (C) , 21.8 (Me), 21.6 (Me), 21.3 (Me). In particular, with the help of an HMBC experiment, the ¹H-NMR spectrum displayed seven Me signals at $\delta(H)$ 0.89 (s, Me(29), Me(30)); 1.02 (s, Me(23)); 1.05 (s, Me(25)); 1.09 (s, Me(24)); 1.16 (s, Me(26)); 1.33 (s, Me(27)), an olefinic signal at $\delta(H)$ 5.34 (br. s, H-C(12)), two oxygenated CH groups at $\delta(H)$ 5.47 (dd, J = 11.4, 5.4, CH), 5.12 (br. d, J = 11.4, CH), and an oxygenated CH₂ group at $\delta(H)$ 4.07 (d, J = 10.4, H_a – C(28)); 4.01 (d, J = 10.4, $H_b-C(28)$; the ¹³C-NMR spectrum showed 30 C-atom signals (7 \times Me, 9 \times CH₂, 6 \times $CH, 8 \times C¹$) for the skeleton of 1a, and additionally three AcO groups, which indicated that compound $1a$ was a pentacyclic triterpene with a trisubstituted C=C bond.

With the aid of HMBC experiments, an oleane-type triterpene skeleton was suggested by the key HMBC correlations of the seven Me groups $(Fig. 2)$, and the correlations of Me(23) and Me(24) with C(3) at δ (C) 216.9 (C) indicated that C(3) is a C=O group; correlation of Me(26) with C(7) at δ (C) 75.2 (CH) indicated that an AcO group was connected with $C(7)$. With the consideration of a chair-like conformation of the five rings of the skeleton of olean-type triterpenes, the coupling pattern of $H-C(7)$ at $\delta(H)$ 5.12 (br. d, J = 11.4, CH) indicated that H-C(7) should be axial and α orientated. The HMBC correlation of the H-atom at $\delta(H)$ 5.47 (dd, J = 11.4, 5.4, CH) with C(28) at δ (C) 66.2 (CH₂) indicated that another AcO group was connected to $C(16)$ or $C(22)$. In an NOE difference experiment (*Fig. 3*), irradiation of the H-atom at

¹⁾ Assigned through a DEPT spectrum.

	$\delta(C)$		$\delta(C)$
CH ₂ (1)	38.9 (t)	$H - C(16)$	68.5(d)
CH ₂ (2)	33.8 (t)	C(17)	39.6 (s)
C(3)	216.9(s)	$H - C(18)$	43.4 (d)
C(4)	47.2 (s)	CH ₂ (19)	46.0 (t)
$H-C(5)$	51.9 (d)	C(20)	30.9(s)
CH ₂ (6)	26.3(t)	CH ₂ (21)	34.1 (t)
$H-C(7)$	75.2 (d)	CH ₂ (22)	23.6(t)
C(8)	44.5 (s)	Me(23)	27.1 (q)
$H-C(9)$	46.3 (d)	Me(24)	21.8 (q)
C(10)	36.8(s)	Me(25)	15.5 (q)
CH ₂ (11)	23.6(t)	Me(26)	11.3 (q)
$H - C(12)$	124.0 (d)	Me(27)	27.0 (q)
C(13)	141.3 (s)	CH ₂ (28)	66.2 (t)
C(14)	44.5 (s)	Me(29)	33.1 (q)
CH ₂ (15)	34.4 (t)	Me(30)	23.6 (q)

Table 1. ¹³C-NMR Data for Compound $1a^a$). δ in ppm.

 $\mathbb{C}1_3$ with Me₄Si as an internal standard. AcO: δ (C) 170.3, 170.3, 171.5, 21.8 (Me), 21.6 (Me), 21.3 (Me).

 $\delta(H)$ 5.47 (dd, J = 11.4, 5.4, CH) produced an NOE enhancement (9.24%) of Me(27), which revealed that the AcO group in question is located at $C(16)$ with β -configuration.

Fig. 2. Key HMBC correlations of compound 1a

The above described analysis of the 1D- and 2D-NMR data, together with a comparative study of NMR data of compound 1a with those of 16,28-dihydroxyolean-12-en-3-one [5], enabled us to elucidate the structure of compound 1a as $7\beta,16\beta,28$ triacetoxyolean-12-en-3-one.

The molecular formula of compound 2a, an optically active oil, could be deduced as $C_{18}H_{24}O_5$ from a HR-ESI-MS experiment with a signal at m/z [M + Na]⁺ 343.1512 (calc. 343.1521), which accounted for seven degrees of unsaturation. The IR absorptions at 1655, 1737, and 1765 cm⁻¹ pointed to the presence of an α , β -unsaturated ketone moiety.

The ¹H-NMR spectrum (*Table 2*) of **2a** showed three Me signals at $\delta(H)$ 2.34 (s), 0.96 (s), and 0.97 (d, $J = 5.7$), the ¹³C-NMR spectrum showed 14 C-atom signals (3 \times Me, $3\times \text{CH}_2, 3\times \text{CH}, 5\times \text{C}^1)$) for the skeleton of $\bm{2a},$ and additionally two AcO groups at $\delta(H)$ 2.26 (s, Me), 2.02 (s, Me), and $\delta(C)$ 170.1 (C), 168.5 (C), 21.7 (Me), 21.4 (Me).

Fig. 3. Key NOE correlation observed for $H-C(16)$ of compound 1a

Table 2. ¹H- and ¹³C-NMR Data for Compounds 2a and 3a. δ in ppm.

	$2a^a$)		$3a^a)$		
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$	
$H-C(1)$	5.48 (br. s)	72.4 (d)	5.44 (br. s)	73.7 (d)	
$H_a - C(2)$	$1.80 - 1.91$ (<i>m</i>)	29.9(t)	$1.95 - 2.00$ (<i>m</i>)	31.1(t)	
$H_b - C(2)$	$1.62 - 1.65$ (<i>m</i>)		$1.65 - 1.71$ (m)		
$H_a - C(3)$	$1.62 - 1.65$ (<i>m</i>)	25.6(t)	$1.65 - 1.71$ (m)	25.5(t)	
$H_b - C(3)$	$1.40 - 1.45$ (<i>m</i>)		$1.40 - 1.50$ (m)		
$H - C(4)$	$1.48 - 1.50$ (<i>m</i>)	43.1 (d)	$1.40 - 1.50$ (<i>m</i>)	42.1 (d)	
C(5)		37.5(s)		40.2(s)	
$H_a-C(6)$	2.92 $(d, J = 12.2)$	25.6(t)	2.81 $(d, J = 14.0)$	39.6 (t)	
$H_b - C(6)$	2.18 $(d, J = 12.2)$		2.10(m)		
C(7)		121.4 (s)		121.6(s)	
C(8)		150.3(s)		190.0 (s)	
$H-C(9)$	5.88 (s)	125.1 (d)	6.02(s)	131.1 (d)	
C(10)		151.4(s)		155.4(s)	
C(11)		196.7 (s)		161.4 (s)	
Me(13)	2.34(s)	31.0 (q)	2.28 $(d, J = 2.4)$	19.0 (q)	
Me(14)	0.96(s)	17.1 (q)	1.07(s)	18.6 (q)	
Me(15)	$0.97 (d, J = 5.7)$	16.1 (q)	0.96 $(d, J = 6.4)$	15.6 (q)	
AcO	2.26(s)	21.7 (q), 170.1 (s)	2.21(s)	21.5 (q), 170.1 (s)	
AcO	2.02(s)	21.4 (q), 168.5 (s)	2.01(s)	21.0 (q), 171.1 (s)	

^a) Recorded in CDCl₃ with Me₄Si as an internal standard.

These findings indicated that compound 2a is an eremophilane derivative containing an α , β -unsaturated ketone moiety [3c].

The spin coupled systems of $H-C(1)$, $CH₂(2)$, $CH₂(3)$, and $H-C(4)$, revealed by a ¹H,¹H-COSY spectrum, and the key HMBC correlations (*Fig. 4*) of the three Me groups of compound 2a confirmed the eremophilane sesquiterpene. The HMBC correlations of Me(13) with C(11) and C(7), and CH₂(6) with C(7), C(11), C(5), C(14), and C(8) showed C(11) as C=O group, correlations of H-C(9) with C(7), $C(1)$, $C(5)$, and $C(8)$, and $Me(13)$ with $C(7)$ pointed to the location of an AcO group at C(1). It also indicated the presence of a tetrasubstituted $\Delta^{7,8}-C=$ C bond and a trisubstituted $\Delta^{9,10}$ -C=C bond. Thus, C(8) must bear an AcO group.

For biogenetic reasons, the Me groups at C(4) and C(5) were both assigned as the β orientation [6]. With the consideration of a chair-like conformation of the A ring of the skeleton of eremophilane sesquiterpenes, the broad *singlet* of $H-C(1)$ at $\delta(H)$ 5.48 (br.

COSY Fig. 4. Key ¹H,¹H-COSY and HMBC correlations of compounds 2a – 4a, 5, and 6 (only substructure shown)**HMBC**

s, CH) with a possible small coupling constant pointed to $H-C(1)$ as being α -orientated [3c] [7].

From the above described analysis of the 1D- and 2D-NMR data, the structure of compound 2a was elucidated as $(1R, 4S, 4aR)$ -6-acetyl-1,2,3,4,4a,5-hexahydro-4,4adimethylnaphthalene-1,7-diyl diacetate.

Compound 3a has the same molecular formula as 2a according to its HR-ESI-MS at m/z $[M+H]^+$ 321.1701 (calc. 321.1702). The IR absorptions at 1633, 1681, and 1739 cm⁻¹ showed the presence of an α , β -unsaturated ketone moiety. The ¹H- and ¹³C-NMR spectral data of compound **3a** (*Table 2*) closely resembled those of compound 2a, indicating that 3a has a close structural relationship to 2a.

The HMBC correlations of Me(13) with C(7) and C(11), and CH₂(6) with C(5), Me(14), C(7), C(8), and C(11), showed C(8) as C=O group, and also indicated the presence of a tetrasubstituted $\Delta^{7,11}$ -C=C bond, which is supported by the evidence that Me(13) at $\delta(H)$ 2.28 (d, $J = 2.4$, Me) showed a homoallylic coupling with H_a-C(6) [7]. HMBC correlations of Me(15) with C(10), and H-C(9) with C(1), C(5), C(7), and $C(8)$ showed the presence of a trisubstituted $\Delta^{9,10}$ -C=C bond; thus, C(11) must bear an AcO group. As in the case of 2a, 3a also contained a β -orientated AcO group at C(1), accounting for the broad *singlet* of H–C(1) at $\delta(H)$ 5.44 (br. s, H–C(1)).

The above described analysis of 1D- and 2D-NMR data further determined the structure of $3a$ as $(1Z)$ -1-[$(5R, 8S, 8aR)$ -5-acetoxy-6,7,8,8a-tetrahydro-8,8a-dimethyl-3oxonaphthalen- $2(1H)$ -ylidene]ethyl acetate.

Compound 4a was obtained as an optically active oil, the molecular formula was determined as $C_{19}H_{24}O_7$ by the signal of a HR-ESI-MS experiment at m/z [$M + Na$]⁺ 387.1413 (calc. 387.1420). The IR spectrum showed absorption bands of OH $(3422 \text{ cm}^{-1}), \text{ C=C } (1675 \text{ cm}^{-1}), \text{ and } \text{C=O } (1780 \text{ and } 1740 \text{ cm}^{-1}).$

The ¹H-NMR spectrum (*Table 3*) of **4a** showed three Me group signals at δ (H) 2.03 (s, Me), 1.18 (s, Me), and 0.87 (d, $J = 5.6$, Me), the ¹³C-NMR spectrum (*Table 3*) showed 15 C-atom signals $(3 \times Me, 2 \times CH_2, 4 \times CH, 6 \times C^1)$ for the skeleton of **4a**, and additionally two AcO groups at $\delta(H)$ 2.17 (s, Me), 2.13 (s, Me), and $\delta(C)$ 170.1 (C) , 169.4 (C) , 21.4 (Me) , 21.1 (Me) . This evidence indicated that compound 4a is an eremophilane derivative with an α , β -unsaturated lactone γ -moiety [8].

In the HMBC spectrum, correlations of Me(14) with $C(4)$, $C(5)$, $C(6)$, and $C(10)$, and of $H-C(6)$ with $C(5)$, C=O of AcO, and C(10) indicated that a OH group is located at $C(10)$ and an AcO group at $C(6)$; correlations of $H-C(1)$ with $C(10)$, $C(9)$, and C=O of AcO indicated an AcO group at $C(1)$; correlations of Me(13) with the tetrasubstituted $\Delta^{7,11}$ -C=C bond, and C(12)=O, H-C(6) with C(7), C(11), and C(8), together with the correlations of $H-C(9)$ with $C(5)$, $C(7)$, and $C(8)$, confirmed the presence of an α , β -unsaturated γ -lactone.

The relative configuration of compound 4a was obtained by a combination of coupling patterns and an NOE difference spectrum, the coupling pattern of $H-C(1)$ at

	$4a^a$)		$(5^b)^c$		6°)	
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$
$H - C(1)$	5.12	73.7 (d)	4.28 (br. s)	70.0 (d)	4.50 (br. s)	73.7 (d)
	$(dd, J = 12.0, 4.0)$					
$H_a - C(2)$	$1.80 - 1.91$ (<i>m</i>)	28.6(t)	$1.95 - 2.00$ (m)	28.5(t)	$1.95 - 2.00(m)$	35.1 (t)
$H_h - C(2)$	$1.62 - 1.65$ (<i>m</i>)		$1.65 - 1.71$ (m)		$1.65 - 1.71$ (m)	
$H_a - C(3)$	$1.62 - 1.65$ (<i>m</i>)	28.5(t)	$1.65 - 1.71$ (<i>m</i>)	28.2(t)	$1.65 - 1.71$ (<i>m</i>)	25.3(t)
$H_b - C(3)$	$1.40 - 1.45$ (<i>m</i>)		$1.40 - 1.50$ (<i>m</i>)		$1.40 - 1.50$ (<i>m</i>)	
$H - C(4)$	$1.48 - 1.50$ (<i>m</i>)	34.3 (d)	$2.30 - 2.40$ (<i>m</i>)	36.1 (d)	$1.50 - 1.60$ (<i>m</i>)	41.9 (d)
C(5)		46.4(s)		51.8 (s)		44.0 (s)
$H-C(6)$	5.97 (s)	69.9 (d)	6.67 $(d, J = 2.4)$	73.7 (d)	6.80(s)	153.4(d)
C(7)		142.1 (s)		147.0 (s)		138.8 (s)
C(8)		149.1 (s)	-	136.1(s)	$\qquad \qquad -$	187.9(s)
$H_a - C(9)$	6.04(s)	107.1 (d)	$2.00 - 2.10$ (<i>m</i>)	36.1 (t)	6.14(s)	126.1 (d)
$H_b - C(9)$			$1.80 - 1.90$ (<i>m</i>)			
C(10)		76.1(s)		80.3(s)		166.4 (s)
C(11)		128.5 (s)		122.0 (s)	3.00(m)	34.7 (d)
$H_a - C(12)$		170.3(s)	7.65 (s)	147.0 (d)	3.57	67.6 (t)
					$(dd, J = 10.8, 5.4)$	
$H_h - C(12)$					3.51	
					$(dd, J = 12.0, 4.0)$	
Me(13)	2.03(s)	9.5 (q)	1.84 (s)	8.7 (q)	1.09 $(d, J = 6.6)$	16.0 (q)
Me(14)	1.18(s)	16.1 (q)	1.21(s)	10.0 (q)	1.31(s)	19.1 (q)
Me(15)	0.87 $(d, J = 5.6)$	11.0 (q)	0.85 $(d, J = 6.3)$	18.0 (q)	1.09 $(d, J = 6.6)$	16.4 (q)
AcO	2.17(s)	21.4 (q),				
		170.1(s)				
AcO	2.13(s)	$21.1 (q)$,				
		169.4(s)				

Table 3. ¹H- and ¹³C-NMR Data for Compound **4a**, **5**, and **6**. δ in ppm.

^a) Recorded in CDCl₃ with Me₄Si as an internal standard. ^b) Recorded in acetone with Me₄Si as an internal standard. \degree) hydroxymethylacryloyl substituent: $\delta(H)$ 6.35 (d, $J = 1.8$, $H_a - C(3)$); 6.06 (d, $J =$ 2.4, $H_b-C(3)$; 4.36 (s, CH₂(4)); $\delta(C)$ 166.6 (C(1)); 142.1 (C (2)); 125.5 (CH₂(3)); 60.9 (CH₂(4)).

 $\delta(H)$ 5.12 (dd, J = 12.0, 4.0, CH) revealed the β -orientation of H - C(1), which was further confirmed by the evidence that irradiation of $H-C(1)$ produced an NOE enhancement of Me(15) (2.81%); irridiation of Me(14) produced an NOE enhancement of H-C(6) (1.82%), indicating that H-C(6) is β -orientated, OH-C(10) was determined to be β -orientated with the consideration of the relative low-field chemical shift of Me(14) in relation to Me(15) [9]. A comparative study of the NMR data of compound 4a (Table 3) with those of 10β -hydroxyeremophila-8(9),7(11)-diene- $6a,15;8,12$ -diolide [10] confirmed the above result.

The above described analysis of the 1D- and 2D-NMR data further determined the structure of $\bf{4a}$ as $(4R, 4aS, 5S, 8s, 8aS)$ -2,4,4a,5,6,7,8,8a-octahydro-8a-hydroxy-3,4a,5trimethyl-2-oxonaphtho[2,3-b]furan-4,8-diyl diacetate.

Compound 5 was a white amorphous powder, the EI-MS gave a peak of M^+ at m/z 350. Together with the analysis of ¹ H-NMR, 13C-NMR, and DEPT spectra, the molecular formula could be deduced as $C_{19}H_{26}O_6$. The IR spectrum showed the

absorption bands of OH (3410 cm⁻¹), an α , β -unsaturated ester (1629, 1699, 1740, and 1780 cm^{-1}), and a furane moiety (1161 and 1010 cm⁻¹).

The ¹H-NMR spectrum (*Table 3*) of 5 showed three Me groups for an eremophilane sesquiterpene at $\delta(H)$ 1.84 (s), 1.21 (s), and 0.85 (d, $J = 6.3$), a hydroxymethylacryloyl group could be distinguished by a typical set of signals at $\delta(H)$ 6.35 (d, J = 1.8, CH), 6.06 (d, $J = 2.4$, CH), 4.36 (s, CH₂), and δ (C) 166.6 (C), 142.1 (C), 125.5 (CH₂), 60.9 $(CH₂)$, which is supported by the fragment ion peak at m/z 85.0 ($[C₄H₅O₂]+$). Another set of signals at $\delta(H)$ 7.65 (s, CH), and $\delta(C)$ 147.0 (CH), 147.0 (C), 136.1 (C), 122.0 (C) revealed the presence of a trisubstituted furan moiety.

The HMBC correlations of Me(14) with C(5), C(6), and C(10), and of $H-C(6)$ with $C(1')$ revealed the position of the OH group at $C(10)$, and the hydroxymethylacryloyl group at $C(6)$; correlations of Me(13) with $C(12)$, $C(11)$, and $C(7)$, of H-C(6) with $C(5)$, $C(8)$, and $C(11)$, and of $H-C(12)$ with $C(8)$, $C(11)$, and $C(7)$, confirmed the presence of a furan moiety. The broad *singlet* of $H-C(1)$ at $\delta(H)$ 4.28 (br. s, CH) showed the relative configuration of $H-C(1)$ as α -orientated. Irradiation of $H-C(6)$ produced an NOE enhancement of H-C(4) (1.82%), indicating that H-C(6) was α orientated. OH – C(10) was determined to be β -orientated with the consideration of the relatively low-field chemical shift of Me(14) in relation to Me(15) [9] [11a], and a comparative study of NMR data of compound 5 (*Table 3*) with those of known compounds [11].

The above described analysis of the 1D- and 2D-NMR data further determined the structure of 5 as (4S,4aS,5S,8R,8aS)-4,4a,5,6,7,8,8a,9-octahydro-8,8a-dihydroxy-3,4a,5 trimethylnaphtho[2,3-b]furan-4-yl (2-(hydroxymethyl)prop-2-enoate.

Compound 6 was obtained as a white amorphous powder. The molecular-ion peak was observed at m/z 250 in the EI-MS, the formula corresponding to this ion peak was determined as $\rm C_{15}H_{22}O_3$ by combined spectroscopic methods (¹H-NMR, ¹³C-NMR, and DEPT). The IR spectrum of 6 indicated the presence of OH (3433 cm^{-1}) , and an $\alpha, \beta, \alpha', \beta'$ -unsaturated ketone moiety (1629, 1664, and 1699 cm⁻¹). A comparative study revealed that the NMR data of compound 6 (*Table 3*) was very similar to those of related known compounds [12].

The ¹H-NMR spectrum of 6 displayed three Me groups for an eremophilane sesquiterpene at $\delta(H)$ 1.31 (s, Me), 1.09 (d, $J = 6.6$, $2 \times$ Me), two olefinic signals at $\delta(H)$ 6.80 (s, CH), and 6.14 (s, CH) for two pairs of trisubstituted C=C bonds, it also showed one oxygenated CH group at $\delta(H)$ 4.50 (br. s, CH) and one oxygenated CH₂ group at $\delta(H)$ 3.57 (dd, J = 10.8, 5.4, CH), 3.51 (dd, J = 12.0, 4.0, CH). The ¹³C-NMR spectrum showed 15 C-atom signals $(3 \times Me, 3 \times CH, 2)$ (one oxygenated), $5 \times CH$ (one oxygenated), $4 \times C$) with the aid of DEPT experiments.

The ¹H,¹H-COSY spectrum showed two spin coupling systems $a(H-C(1), CH₂(2),$ $CH₂(3)$, and $H-C(4)$) and b (CH₂(12), H-C(11), and Me(13)). In the HMBC spectrum, the correlations of Me(14) with $C(5)$, $C(6)$, and $C(10)$, and Me(15) with $C(4)$ and $C(5)$ indicated the presence of two pairs of C=C bonds at $C(6)$ and $C(10)$, correlations of Me(13) with C(11), C(7), and C(12), and of H-C(12) with C(7), together with the correlations of $H-C(6)$ with $C(8)$, and $C(10)$, and of $H-C(9)$ with C(7) further confirmed the C=C bonds as $C(6)=C(7)$ and $C(9)=C(10)$, and also revealed the presence of a C=O group at C(8). The position of the OH group at C(1) was deduced from the HMBC correlations of $H-C(9)$ with C(1), and of $H-C(1)$ with C(9) and C(5). The relative configuration of H-C(1) was determined to be α by the broad *singlet* at $\delta(H)$ 4.50 (br. s, H-C(1)).

From the above discussed data, the structure of 6 was finally determined as (4aS,5S,8R)-5,6,7,8-tetrahydro-8-hydroxy-3-(1-hydroxypropan-2-yl)-4a,5-dimethylnaphthalen-2(4aH)-one by analysis of 1D- and 2D-NMR data.

Compound 7 was obtained as a white amorphous powder. The molecular formula was determined as $C_{12}H_{16}O_3$ by a HR-ESI-MS experiment at m/z $[M+H]^+$ 209.1170 (calc. 209.1178). The IR spectrum of 7 indicated the presence of OH (3443 cm⁻¹), and an $\alpha, \beta, \alpha', \beta'$ -unsaturated ketone moiety (1629, 1686 cm⁻¹). A comparative study revealed that the NMR data of compound 7 (Table 4) was similar to those of a known related compound [3c].

	7°)		$8^a)$		
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$	
$H-C(1)$	4.59 (br. s)	74.2 (d)	4.36 (br. s)	73.0 (d)	
$H_a - C(2)$	$2.05 - 2.32$ (<i>m</i>)	35.0(t)	$2.05 - 2.32$ (<i>m</i>)	33.1 (t)	
$H_h - C(2)$	$1.60 - 1.70$ (<i>m</i>)		$1.60 - 1.70$ (<i>m</i>)		
$H_a - C(3)$	$1.80 - 1.90$ (<i>m</i>)	25.0(t)	$1.80 - 1.90$ (<i>m</i>)	30.0 (t)	
$H_b - C(3)$	$1.40 - 1.50$ (<i>m</i>)		$1.40 - 1.50$ (<i>m</i>)		
$H - C(4)$	$1.50 - 1.60$ (<i>m</i>)	43.0 (d)	$1.50 - 1.60$ (<i>m</i>)	43.8 (d)	
C(5)		44.7 (s)		40.8 (s)	
$H_a - C(6)$	6.33(s)	126.7(d)	2.21 $(dd, J=12.3, 5.4)$	43.1 (t)	
$Hb-C(6)$			2.00 (dd, $J = 12.3, 3.0$)		
$H - C(7)$		146.0 (s)	5.56 (dd, $J = 14.1, 5.4$)	71.7(s)	
C(8)		182.4 (s)		194.9 (s)	
$H-C(9)$	6.28(s)	123.7(d)	5.86 (s)	124.8 (d)	
C(10)		169.4(s)		170.5(s)	
Me(14)	1.38(s)	19.6 (q)	1.44 (s)	15.2 (q)	
Me(15)	1.10 $(d, J=6.3)$	16.7 (q)	0.94 $(d, J=6.9)$	19.0 (q)	
OН	6.20 (s)				
AcO			2.17(s)	21.1 (q),	
				170.5(s)	

Table 4. ¹H- and ¹³C-NMR Data for Compound **7** and 8. δ in ppm.

^a) Recorded in CDCl₃ with Me₄Si as an internal standard.

The ¹H-NMR spectrum of 7 displayed two Me groups at $\delta(H)$ 1.38 (s) and 1.10 (d, $J = 6.3$), two olefinic signals at $\delta(H)$ 6.33 (s, CH) and 6.28 (s, CH) for two pairs of trisubstituted C=C bonds, and an oxygenated CH at $\delta(H)$ 4.59 (br. s, CH). The ¹³C-NMR spectrum showed 12 C-atom signals $(2 \times Me, 2 \times CH_2, 4 \times CH)$ (one oxygenated), $4 \times C^1$) for the skeleton of 7.

The key ${}^{1}H, {}^{1}H$ -COSY and HMBC correlations (*Fig. 5*) showed that compound 7 was a tri-nor eremophilane sesquiterpene. In the HMBC spectrum, correlations of $H-C(6)$ with $C(5)$, $C(7)$, $C(8)$, $C(10)$, and $Me(14)$, and of $H-C(9)$ with $C(7)$, $C(1)$, and C(5), confirmed the presence of an $\alpha, \beta, \alpha', \beta'$ -unsaturated ketone moiety. Correlations of the H-atom of a OH group at $\delta(H)$ 6.20 (s) with C(6), C(7), and $C(8)$ confirmed the location of OH at $C(7)$, correlations of H-C(9) with C(1), and $H-C(1)$ with $C(9)$ and $C(5)$ indicated that another OH group was located at $C(1)$.

The above discussed analysis of the 1D- and 2D-NMR data of compound 7 further showed the structure of 7 as (4aS,5S,8R)-5,6,7,8-tetrahydro-3,8-dihydroxy-4a,5-dimethylnaphthalen- $2(4aH)$ -one.

Compound 8 was obtained as a white amorphous powder. The HR-ESI-MS experiment showed the molecular formula as $C_{14}H_{20}O_4$ at m/z $[M + Na]^+$ 275.1259 (calc. 275.1259). The IR spectrum of 8 indicated the presence of OH (3410 cm⁻¹), and an α , β -unsaturated ketone moiety (1736 and 1656 cm⁻¹). The NMR data of compound 8 (Table 4) was similar to those of 7.

The ¹H-NMR spectrum of 8 displayed two Me groups at $\delta(H)$ 1.44 (s) and 0.94 (d, $J = 6.9$), one olefinic signal at $\delta(H)$ 5.86 (s, CH), and two oxygenated CH groups at $\delta(H)$ 5.56 (dd, J = 14.1, 5.4, CH) and 4.36 (br. s, CH). The ¹³C-NMR showed 12 C-atom signals (2 \times Me, 3 \times CH₂, 4 \times CH (two oxygenated), 3 \times C¹)) for the skeleton of **8**, and additionally two signals for an AcO group (δ (H) 2.17 (s, Me), δ (C) 170.5 (C) and 21.1 (Me)).

Compound 8 was also determined as a tri-nor eremophilane sesquiterpene with the aid of the key ${}^{1}H, {}^{1}H$ -COSY and HMBC correlations (*Fig. 5*). The HMBC correlations of CH₂(6) with C(8), C(7), C(5), and Me(14), together with the crosspeak of H-C(7)/ $CH₂(6)$ in the ¹H,¹H-COSY spectrum, indicated the presence of an AcO group at C(7). $H-C(1)$ was deduced to be α -orientated due to the coupling pattern of a broad *singlet* at $\delta(H)$ 4.36 (br. s, CH), irradiation of H-C(7) produced an NOE enhancement of Me(14) (2.01%), indicating that H-C(7) must be β -orientated.

The above described analysis of the 1D- and 2D-NMR data further determined the structure of $\boldsymbol{8}$ as $(2R, 5R, 8S, 8aR)$ -1,2,3,5,6,7,8,8a-octahydro-5-hydroxy-8,8a-dimethyl-3oxonaphthalen-2-yl acetate.

2. Biological Studies. Compound 1a was tested for its cytotoxic activity against HL-60 (human promyelocytic leukemia), SMMC-7721 (human hepatoma), and HeLa (human cervical carcinoma) cells according to the sulforhodamine B (SRB) method [13] (use of vincristine sulfate as a positive control with IC_{50} values 11.2 µg/ml against HL-60, 26.7 µg/ml against SMMC-7721, and 8.3 µg/ml against HeLa) as reported previously. Compound 1a showed moderate cytotoxicity against HL-60 and SMMC-7721 cells with IC_{50} values of 11.2 and 42.6 µg/ml, respectively, while it was not bioactive against HeLa cells with an IC_{50} value of more than 100 µg/ml.

Experimental Part

General. TLC: $GF_{254} (10-40 \,\mu)$; spots were detected under UV or by heating after spraying with 5% $H₂SO₄$ in EtOH (v/v). Column chromatography (CC): silica gel (SiO₂; 200 – 300 mesh) was supplied by the *Oingdao Marine Chemical Factory* in P. R. China. The reversed phase pre-coated TLC plates: RP-18 F_{254} (size 20 \times 20 cm, Schichtdicke 0.25 mm); supplied by E. Merck, Germany. M.p.: Kofler melting-point apparatus, uncorrected. Optical rotations: Perkin Elmer 341 polarimeter. IR: Nicolet NEXUS 670 FT-IR; in cm⁻¹. NMR: Varian Mercury plus-300 NMR spectrometer; with Me₄Si as an internal standard. EI-MS: HP-5988A GC/MS instrument; m/z (rel. int.). HR-ESI-MS: Bruker Daltonics APEX II 47e spectrometer.

Plant Material. The roots of L. sagitta MAXIM were collected from Gannan Tibet Autonomous Region (S. A. 2000 – 3800 m), Gansu Province of P. R. China in August 2005. It was identified by Prof. Guo-Liang Zhang, School of Life Sciences, Lanzhou University. A voucher specimen (No. 20050920) was deposited with the College of Chemistry and Chemical Engineering, Lanzhou University.

Extraction and Isolation. The air-dried roots of L. sagitta (10 kg) were pulverized, and extracted with mixed solvent (petroleum ether (PE) (b.p. $60 - 90^{\circ}$)/Et₂O/MeOH 1:1:1) three times (7 d each time) at r.t. The extract was concentrated under reduced pressure, the residue (400 g) was subjected to $SiO₂$ CC, and eluted with a step gradient of PE/acetone (20:1, 10:1, 5:1, 3:1, and 1:1). Five fractions (Fr. 1 – 5) were collected according to TLC. Fr. 3 (PE/acetone 5:1; 50 g) was purified by repeated chromatography over a SiO₂ column with PE/acetone (5:1) to afford $7(5 \text{ mg})$, $8(3 \text{ mg})$, and a mixture. After acetylation of the mixture, 2a (5 mg) and 3a (3 mg) were obtained. Fr. 4 (PE/acetone 3:1; 40 g) after SiO₂ CC (200 – 300 mesh; 400 g) with PE/acetone (3:1) as eluent, three mixtures were obtained. Two of them were then rechromatographed by $RP-18$ PTLC (H₂O/MeOH 1:3, two times each) to afford $5(2 \text{ mg})$ and $6(13 \text{ mg})$, resp., and the third one was acetylated to afford **1a** (2 mg) and **4a** (3 mg) .

(7 β ,16 β)-3-Oxoolean-12-ene-7,16,28-triyl Triacetate (1a). White amorphous powder. M.p. 190 – 192°. $\lbrack \alpha \rbrack_{D}^{20} = +4.0 \text{ } (c = 0.2, \text{CHCl}_3)$. IR (film): 2927, 2855, 1737, 1706, 1634, 1459, 1372, 1242, 1021. ¹H-NMR $(300 \text{ MHz}, \text{CDCl}_3): 0.89 \text{ (s, Me(29), Me(30))}; 1.02 \text{ (s, Me(23))}; 1.05 \text{ (s, Me(25))}; 1.09 \text{ (s, Me(24))}; 1.16 \text{ (s,$ $Me(26)$; 1.33 (s, Me(27)); 1.94 (s, AcO); 2.01 (s, AcO); 2.09 (s, AcO); 4.01 (d, J = 10.4, H_b-C(28)); 4.07 $(d, J = 10.4, H_a-C(28))$; 5.12 (br. d, J = 11.4, H – C(7)); 5.34 (br. s, H – C(12)); 5.47 (dd, J = 11.4, 5.4, $\text{H}-\text{C}(16)$). ¹³C-NMR: *Table 1*. HR-ESI-MS: 616.4212 ([$M+\text{NH}_4$]⁺, C₃₆H₅₈NO₇⁺; calc. 616.4213).

(1R,4S,4aR)-6-Acetyl-1,2,3,4,4a,5-hexahydro-4,4a-dimethylnaphthalene-1,7-diyl Diacetate (2a). Optically active oil. $\lbrack \alpha \rbrack_{D}^{20} = +26$ (c = 0.5, CHCl₃). IR (film): 2959, 2931, 2865, 1765, 1737, 1655, 1595, 1370, 1238, 1189, 1016. ¹H- and ¹³C-NMR: *Table 2*. HR-ESI-MS: 343.1512 ($[M + Na]^+$, $C_{18}H_{24}NaO_5^+$; calc. 343.1521).

(1Z)-1-[(5R,8S,8aR)-5-(Acetyloxy)-3,5,6,7,8,8a-hexahydro-8,8a-dimethyl-3-oxonaphthalen-2(1H) *ylidene]ethyl Acetate* (3a). Optically active oil. $[a]_D^{20} = +14$ ($c = 0.3$, CHCl₃). IR (film): 2928, 2856, 1758, 1739, 1681, 1633, 1371, 1255, 1180, 1015. 'H- and ¹³C-NMR: *Table 2*. HR-ESI-MS: 321.1701 ($[M+H]^+$, $C_{18}H_{25}O_5^+$; calc. 321.1702).

(4R,4aS,5S,8S,8aS)-2,4,4a,5,6,7,8,8a-Octahydro-8a-hydroxy-3,4a,5-trimethyl-2-oxonaphtho[2,3 b]*furan-4*,8-diyl Diacetate (4a). Optically active oil. $\lbrack a \rbrack_{0}^{\infty} = -3$ ($c = 0.3$, CHCl₃). IR (film): 3422, 2922, 2853, 1780, 1740, 1675, 1376, 1235, 1039, 1021. ¹H- and ¹³C-NMR: *Table 3.* HR-ESI-MS: 387.1413 ([*M* + Na]⁺, C₁₉H₂₄NaO₇⁺; calc. 387.1420).

(4S,4aS,5S,8R,8aS)-4,4a,5,6,7,8,8a,9-Octahydro-8,8a-dihydroxy-3,4a,5-trimethylnaphtho[2,3-b]furan-4-yl 2-(Hydroxymethyl)prop-2-enoate (5). White amorphous powder. M.p. $185-186^{\circ}$. [α] $_{10}^{120}$ = -10 (c = 0.2, CHCl3). IR (film): 3410, 2927, 2856, 1780, 1740, 1699, 1663, 1629, 1367, 1215, 1161, 1099, 1061, 1010. ¹H- and ¹³C-NMR: *Table 3*. EI-MS: 350 (5, M⁺), 289 (100), 280 (65), 262 (98), 244 (28), 215 (30), 177 (85), 109 (20), 85 (80).

(4aS,5S,8R)-5,6,7,8-Tetrahydro-8-hydroxy-3-(1-hydroxypropan-2-yl)-4a,5-dimethylnaphthalen-2(4aH)-one (6). White amorphous powder. M.p. $190-192^{\circ}$. $\left[\alpha\right]_{10}^{20} = -1.7$ (c = 1.3, CHCl₃). IR (film): 3433, 2928, 2853, 1739, 1699, 1664, 1629, 1467, 1234, 1208, 1022. ¹ H- and 13C-NMR: Table 3. EI-MS: 250 $(10, M⁺), 232 (10), 221 (100), 175 (60), 147(60), 135 (70), 91 (60).$

(4aS,5S,8R)-5,6,7,8-Tetrahydro-3,8-dihydroxy-4a,5-dimethylnaphthalen-2(4aH)-one (7). White amorphous powder. M.p. $187-189^\circ$. $\left[\alpha\right]_0^{20} = -1.4$ ($c = 0.5$, CHCl₃). IR (film): 3443, 2934, 2857, 1759,

1738, 1686, 1629, 1367, 1239, 1181, 1016. ¹H- and ¹³C-NMR: *Table 4*. HR-ESI-MS: 209.1170 ([$M + H$]⁺, $C_{12}H_{17}O_3^+$; calc. 209.1178).

(2R,5R,8S,8aR)-1,2,3,5,6,7,8,8a-Octahydro-5-hydroxy-8,8a-dimethyl-3-oxonaphthalen-2-yl Acetate (8). White amorphous powder. M.p. $188 - 191^{\circ}$. $\left[\alpha\right]_{D}^{20} = +3.0$ ($c = 0.3$, CHCl₃). IR (film): 3410, 2926, 2856, 1756, 1736, 1656, 1373, 1236, 1161, 1015. ¹H- and ¹³C-NMR: *Table 4.* HR-ESI-MS: 275.1259 ([M + Na]⁺, C₁₄H₂₀NaO₄⁺; calc. 275.1259).

Cytotoxicity Assay. Testing for in vitro cytotoxic activities against SMMC-7721 (human hepatoma), HL-60 (human promyelocytic leukemia), and HeLa (human cervical carcinoma) cells of compound 1a was carried out according to the sulforhodamine B (SRB) method [11]. Vincristine sulfate was used as a positive control. The absorbency of extracted sulforhodamine B at 515 nm was measured on a microplate reader. The experiments were carried out in triplicate. Each run entailed 5-6 concentrations of the compounds being tested.

The percentage survival rate of cells exposed to the compound was calculated by assuming the survival rate of untreated cells to be 100%. Plotting the compound concentrations vs. the growth rates of cells provided the half inhibitory concentration IC_{50} values of the compound.

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